

SECONDARY METABOLITES FROM *Voluntaria crupinoides*

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As a part of our ongoing program of research on Algerian Asteraceae [1–4], we report our results on *Voluntaria crupinoides* (Desf.) Maire (Asteraceae), an endemic species to the Sahara [5]. This work concerned the phytochemical study of the chloroform and *n*-butanol soluble parts of the aqueous-ethanol extract of the leaves and flowers of *Voluntaria crupinoides*.

Voluntaria crupinoides was collected during the flowering stage in May 1998 in the Algerian Sahara and was authenticated by Pr. A. Kaabeche (University of Setif Algeria).

Air-dried leaves and flowers (1576 g) of *Voluntaria crupinoides* were macerated in an EtOH solution (70%) for 24 hours three times. After filtration, the filtrate was concentrated (500 mL) and dissolved in H₂O (600 mL). The resulting solution was extracted successively with CHCl₃ and *n*-butanol. The organic phases were dried with Na₂SO₄ filtered, and concentrated under low pressure at room temperature to obtain the extracts: chloroform (12.4 g) and *n*-butanol (25.4 g).

The chloroform extract (6 g) was chromatographed on a silica gel (70–230 mesh) column eluted with a gradient of hexane–EtOAc to yield 12 fractions (F₁–F₁₂). Fraction F₆ (*n*-hexane–EtOAc 90:10) was subjected to preparative TLC on silica gel (*n*-hexane–EtOAc 50:50) to give compound **1** (55 mg).

The *n*-BuOH extract (14 g) was applied to a polyamid SC6 column eluted with a gradient of toluene–MeOH to yield 15 fractions (F₁–F₁₅). Fraction F₆ (toluene–MeOH 80:20) was purified by preparative TLC on polyamide (toluene–MeOH–methyl ethyl ketone 4:3:3) to give compound **2**. Fraction F₉ (toluene–MeOH 70:30) was submitted to preparative TLC on polyamide (water–MeOH–methyl ethyl ketone–acetylacetone 13:3:3:1) followed by purification over Sephadex LH-20 column using MeOH as eluent to give compounds **3** and **4**.

The structures were established by chemical and spectral analysis, mainly FAB-MS (+), IR, UV, ¹H NMR, ¹³C NMR, and NMR experiments (DEPT, COSY, HSQC, and HMBC). All these data were in good agreement with the respective literature data [6–11].

Compound 1. C₁₉H₂₂O₆, IR spectrum (KBr, ν, cm⁻¹): 1751 (C=O of γ-lactone), 1730 (C=O of ester), 3418 (OH). Mass spectrum (FAB⁺, 70 eV), *m/z* : 369 [M + Na]⁺, 347 [MH]⁺.

¹H NMR (250 MHz, CDCl₃, δ, ppm, J/Hz): 2.98 (1H, m, H-1α), 1.74 (1H, m, H-2α), 2.24 (1H, m, H-2β), 4.56 (1H, m, H-3α), 2.85 (1H, t, J_{5α-6β} = J_{5α-1α} = 10.3, H-5α), 4.27 (1H, dd, J_{6β-5α} = 10.3, J_{6β-7α} = 9.2, H-6β), 3.20 (1H, m, H-7α), 5.14 (1H, m, H-8β), 2.72 (1H, dd, J_{9α-9β} = 14.6, J_{9α-8β} = 5.1, H-9α), 2.40 (1H, dd, J_{9β-9α} = 14.6, J_{9β-8β} = 3.7, H-9β), 6.24 (1H, d, J_{13-7α} = 3.4, H-13), 5.63 (1H, d, J_{13'-7α} = 3.0, H-13'), 5.15 (1H, s, H-14), 4.94 (1H, s, H-14'), 5.49 (1H, s, H-15), 5.37 (1H, s, H-15'), 6.34 (1H, s, H-18), 5.97 (1H, s, H-18'), 4.39 (2H, s, H-19).

¹³C NMR (62.9 MHz, CDCl₃, δ, ppm): 45.3 (C-1), 39.0 (C-2), 73.7 (C-3), 152.1 (C-4), 51.5 (C-5), 78.4 (C-6), 47.6 (C-7), 74.3 (C-8), 37.0 (C-9), 141.7 (C-10), 137.3 (C-11), 169.1 (C-12), 122.7 (C-13), 118.1 (C-14), 113.8 (C-15), 165.3 (C-16), 139.2 (C-17), 126.7 (C-18), 62.2 (C-19).

This compound was identified as cynaropicrin [6, 7].

Compound 2. C₁₅H₁₀O₅, mp 349°C. UV (MeOH, λ_{max}, nm): 266, 330; +NaOH: 274, 339, 390; +AlCl₃: 269, 299, 343, 379; +AlCl₃/HCl: 269, 299, 342, 376; +NaOAc: 268, 298, 338; +NaOAc/H₃BO₃: 268, 333.

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¹H NMR (250 MHz, MeOH-d₄, δ, ppm, J/Hz): 7.76 (2H, d, J = 8.5, H-2', H-6'), 6.86 (2H, d, J = 8.5, H-3', H-5'), 6.52 (1H, s, H-3), 6.40 (1H, d, J = 2.1, H-8), 6.12 (1H, d, J = 2.1, H-6).

This compound was characterized as 5,7,4'-trihydroxyflavone (apigenin) [8, 9].

Compound 3. C₁₅H₁₀O₆, mp 330°C. UV (MeOH, λ_{max}, nm): 253, 268, 349; +NaOH: 273, 406; +AlCl₃: 273, 419; +AlCl₃/HCl: 262, 279, 295, 359, 387; +NaOAc: 265, 349; +NaOAc/H₃BO₃: 258, 372.

¹H NMR (250 MHz, MeOH-d₄, δ, ppm, J/Hz): 7.35 (2H, m, H-2', H-6'), 6.90 (1H, d, J = 8.5, H-5'), 6.53 (1H, d, J = 2.0, H-8), 6.42 (1H, s, H-3), 6.15 (1H, d, J = 2.0, H-6).

This compound was characterized as 5,7,3',4'-tetrahydroxyflavone (luteolin) [8, 10].

Compound 4. C₂₂H₂₂O₁₂, UV (MeOH, λ_{max}, nm): 285, 348; +NaOH: 294, 387; +AlCl₃: 276, 302, 388; +AlCl₃/HCl: 284, 317, 384; +NaOAc: 287, 347; +NaOAc/H₃BO₃: 248, 344.

¹H NMR (250 MHz, CDCl₃, δ, ppm, J/Hz): 8.13 (1H, d, J = 2.0, H-2'), 7.97 (1H, dd, J = 8.5, 2.0, H-6'), 7.10 (1H, d, J = 8.5, H-5'), 6.95 (1H, d, J = 2.0, H-8), 6.60 (1H, J = 2.0, H-6), 5.17 (1H, d, J = 7.6, H-1'' glucose), 4.00–3.20 (sugar protons), 4.13 (3H, s, O-CH₃).

Acid hydrolysis of this compound produced isorhamnetin and glucose. This compound was characterized as isorhamnetin 4'-glucoside [8, 11].

REFERENCES

1. Lahcene Zaiter, Mohamed Bouheroum, Samir Benayache, Fadila Benayache, Francisco Leon, Ignacio Brouard, Jose Quintana, Francisco Estevez, and Jaime Bermejo, *Biochem. Syst. Ecol.*, **35**, 533 (2007).
2. Ramdane Seghiri, Ouahiba Boumaza, Ratiba Mekkiou, Samir Benayache, Paul Mosset, Jose Quintana, Francisco Estevez, Francisco Leon, Jaime Bermejo, and Fadila Benayache, *Phytochem. Lett.*, **2**, 114 (2009).
3. L. Aliouche, H. Zater, D. Zama, A. Bentamene, R. Seghiri, R. Mekkiou, S. Benayache, and F. Benayache, *Chem. Nat. Comp.*, **43**, 618 (2007).
4. Azzedine Boudjerda, Hanene Zater, Samir Benayache, Jean-Claude Chalchat, Javier Gonzalez-Platas, Francisco Leon, Ignacio Brouard, Jaime Bermejo, and Fadila Benayache, *Biochem. Syst. Ecol.*, **36**, 461 (2008).
5. P. Quezel and S. Santa, *Nouvelle Flore de l'Algerie et des Regions Desertiques et Meridionales*, Tome **II**, edition CNRS, Paris, 1963.
6. A. Rustaiyan, A. Niknejad, C. Zdero, and F. Bohlmann, *Phytochemistry*, **20**, 2427 (1981).
7. J. A. Marco, J. F. Sanz, R. Albiach, A. Rustaiyan, and Z. Habibi, *Phytochemistry*, **32**, 395 (1993).
8. T. J. Mabry, K. R. Markham, and M. B. Thomas, *The Systematic Identification of Flavonoids*, Springer, New York, 1970, p. 45.
9. E. Wollenweber, *Phytochemistry*, **13**, 753 (1974).
10. C. A. Williams and J. B. Harborne, *Biochem. Syst. Ecol.*, 181 (1975).
11. Y. K. Park and C. Y. Lee, *J. Agric. Food Chem.*, **44**, 34 (1996).